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Balancing Chemical Equations

2 Write formula equations correctly, using proper symbolic notation 3 Correctly balance chemical equations 4 Explain how a balanced chemical equation illustrates the Law of Conservation of Mass (Matter) Considerations: This activity is a wonderful way to introduce the concept of writing and balancing chemical equations

Balancing Chemical Equations Worksheet 1

Balancing Chemical Equations Worksheet 2 - ANSWERS $26 \text{ 2Mg} + \text{Cl}_2 \rightarrow \text{MgCl}_2$ $27 \text{ 2Ag} + \text{O}_2 \rightarrow 4\text{Ag} + \text{O}_2$ $28 \text{ 4K} + 2\text{O}_2 \rightarrow 2\text{K}_2\text{O}$ $29 \text{ Cl}_2 + 3\text{F}_2 \rightarrow 2\text{ClF}_3$ $30 \text{ SiO}_2 + 2\text{C} \rightarrow \text{Si} + 2\text{CO}$ $31 \text{ 2NaHCO}_3 \rightarrow \text{Na}_2\text{CO}_3 + \text{CO}_2 + \text{H}_2\text{O}$ $32 \text{ 2ZnS} + 3\text{O}_2 \rightarrow 2\text{ZnO} + 2\text{SO}_2 \dots$

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