

# Chemical Formulas And Names Word Search Answers

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### Chemical Formulas And Names Word

#### SUMMARY OF RULES FOR CHEMICAL NAMES AND FORMULAS

NAMES: First word is a numerical prefix (omitted if first element occurs only once), then name of first element Second word is a numerical prefix (always present), then the name of the second element changed to end in "-ide" (Second element is always more electronegative, closer to F on the periodic table) FORMULAS: Just write

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#### Chemical Bonding: Names and Formulas

Chemical Names & Formulas Water Ammonia Methane 2 Why "Systematic Names" ? # atomic particles 3 (p, n, e) # elements 110+ # elements in earth's crust (99%) 8 # elements in all living things 25 # compounds >14,000,000 3 Why "Systematic Names" ? Water H 2 O Lime CaO Lye NaOH Potash K 2 CO 3 Table Salt NaCl Laughing Gas N 2 O

#### Writing Chemical Formulas and Chemical Reactions

These elements are NOT binary chemical compounds, but you should know the names and formulas of these elements Use your Periodic Table to determine the noble gas that is at the end of each period period name of gas formula or symbol period 1 period 2 period 3 period 4 period 5 period 6

## 6 NAMING BINARY IONIC COMPOUNDS

**Naming Compounds Handout**

CHEMICAL FORMULAS AND NAMES FROM INDIVIDUAL IONS Compounds are named from the individual ions they come from Name the cation and the anion, then remove "ion" from each name: eg Na + = sodium ion Cl - = chloride ion ~ NaCl = sodium chloride K + = potassium ion Fe +3 = iron (III) ion - Ag + = silver ion S -2 = sulfide ion ~ Ag<sub>2</sub>S

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**Translating Word Equations To Chemical Formulas PDF**

Formulas Summary Of : Translating Word Equations To Chemical Formulas Apr 26, 2020 ~~ eBook Translating Word Equations To Chemical Formulas ~~ By Robin Cook, step 1 identify reactants and products and place them in a word equation sodium hydroxide ironii chloride sodium chloride ironii hydroxide step 2 convert the chemical names into chemical

**Chemical Conversions and Problems**

Conversions between moles and grams of a chemical can be done by using molar mass, which is the addition of the atomic masses of the atoms making up that molecule Atomic mass is the mass in grams of 1 mole, 6022 x 10<sup>23</sup> atoms, of an element These atomic masses are listed on most periodic tables

**Writing Chemical formula with polyatomic groups**

Eg The chemical formula for magnesium nitrate is Mg(NO<sub>3</sub>)<sub>2</sub> and not Mg<sub>2</sub>(NO<sub>3</sub>)<sub>2</sub> or Mg(NO<sub>3</sub>)<sub>2</sub> Rule 2: Brackets() are needed when a polyatomic group > 1 in a compound Poly = 'many' The hydroxide ion, formula OH<sup>-</sup>, is a polyatomic group since it is made up of two different elements Eg The chemical formula for strontium hydroxide is Sr

**"Chemical Names and Formulas"**

Writing Ionic Compound Formulas Example: Ammonium sulfate (note the 2 word name) 1 Write the formulas for the cation and anion, including CHARGES! NH<sub>4</sub><sup>+</sup> + SO<sub>4</sub><sup>2-</sup> 2 Check to see if charges are balanced 3 Balance charges , if necessary, using subscripts Use parentheses if you need more than one of a polyatomic ion Use the criss-cross

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