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Lab: Buffer Preparation and Testing - Travel Pics from ...

Advanced Chemistry with Vernier 19 - 1 Lab: Buffer Preparation and Testing A buffer is a mixture of a weak acid and its conjugate base, or a weak base and its conjugate acid A buffer's function is to absorb acids (H^+ or H_3O^+ ions) or bases (OH^- ions) so that the pH of the system changes very, very little In many systems, buffers are critical

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pH and Buffers Laboratory

buffer solution of known pH and potential differences are read directly in units of pH Overview of the Lab Exercise First you will learn about the general operating techniques used with a pH meter and calibrate the meter at pH 10 Then a 20 ml sample of Na_3PO_4 will be titrated after setting up the burette, the stirrer and the electrode